

## Note Taking Guide Episode 604 Name

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*Note Taking Guide Episode 604 - Part 1*

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*Physics 604: Machines and Mechanical Energy | Georgia ...*

Assuming the pressure stays constant, at what temperature will the volume of the gas be 604 cm<sup>3</sup>? Given: V1 = 506 cm<sup>3</sup>. T1 = 147 °C. V2 = 604 cm<sup>3</sup>. T2 = ? V1/T1 = V2/T2. 506 cm<sup>3</sup>/420 K = 604 cm<sup>3</sup>/T2. T2 = 501 K . or. 501 - 273 = 228 °C. Remember to convert temperature to Kelvin (420 K) Episode 902

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